

Crane Environmental, Inc.

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Crane Water University

Basic Water Treatment for Power Engineering

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Crane Water University Modules





Basic Water Treatment for Power Engineering

- Water Sources
- Basic Water Chemistry
- Common Impurities & Countermeasures
- Water Treatment Technologies Overview
- Basic Reverse Osmosis (RO)
- RO System Design
- Deaeration (DA)
- Boiler Feedwater Treatment
- Cooling Tower Feedwater and Blowdown Water Treatment
- Crane Environmental (CE) Capabilities

The Steam Loop



Natural Water Sources



Cloud Formation Rain Clouds Evaportation Foon keep From Soil From Ocea Precipitation Surface Water X X X X X Infiltration Soil Bedrock apled from: Stream Corridor Restoration Percolation Groundwater Ocean Seawater Deep Percolation

Primary Sources:

 Surface Water: Lakes, Ponds, Rivers, Canals The Hydrologic Cycle

- Ground Water: Wells, Springs
- Seawater: Ocean Intakes

Other Sources:

- Municipal: Town, City, County
- Brackish: Salt Water
 Intrusion
- Process Water

Typical Water Contaminants



Surface

- Typically high in suspended solids & sediment; Low in TDS (< 1000 mg/l)
- Likely to contain organic matter from fallen leaves, aquatic life, etc.
- Highly variable season to season

Ground

- Typically Low in suspended solids & sediment
- Can be high in TDS (>1000 mg/l)
- Likely to contain iron, sulfur, high levels of hardness (region dependent)

Municipal

- Usually has been treated to remove sediment and iron
- May be high in hardness and TDS usually < 500 mg/l
- Most likely will contain chlorine to control biological growth

Brackish & Ocean

- Very high in dissolved solids (TDS from 10,000-40,000 mg/l)
- Mostly sodium and chloride
- Likely to contain organic matter

Basic Water Chemistry



Why is Some Knowledge of Water Chemistry Very Important?

- In order to solve a water need, you first must know what contaminants are present in your supply *Get Complete Feed Water Analysis!!!*
- What levels of these contaminants are required in the treated product? *Process Water Spec? / Concentrations Limits?*
- Understand how the contaminants are measured *TDS*: *PPM / mg/l Conductivity*: μmhos / μS, *Resistivity*: Megohms; *TSS*: NTU / SDI
- Understand how to treat the water to remove specific groups of impurities **Pre-Treat / R.O. / Post-Treat...**



The Water Train



Industrial Water Trains

- A water treatment train is a collection of individual water treatment technologies working as a system to remove contaminants and condition water to meet specific downstream requirements
- The selection of specific components is driven by the characteristics of the incoming feed water against capital equipment and operational costs, size constraints, chemical usage, etc.
- The "ideal" configuration (technology + position in the train) is specific to each application and frequently each site



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Multi-Media Filtration (depth filters)





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Multi-Media Filtration (depth filters)



- Contains several layers of various density gravel, garnet, sand & anthracite
- Used for sediment removal (TSS) and large solids removal
- Requires backwashing
- Effectiveness increases over time
- Does not reduce total dissolved solids (TDS)





- Multi-media filtration has various media layers that trap large particles, and successively smaller particles are trapped in the fine, upper layer of media over time.
- The result is a highly efficient filtering mechanism since removal can take place throughout the entire bed
- Multi-Media filters typically remove particles 10 15 microns in size or larger
- Automatic backwashing steps remove the contaminants trapped within the filter bed and washes them down the drain

Multi-Media (continued)



- Modular design
- Fast, inexpensive installation
- Minimum energy requirements
- Optional PLC Controls
- Backwash based on differential pressure, time or throughput
- Stand Alone or Skid-Mounted





Multi-Media Filtration Alternatives

Screen Filters

- Ideal for very high levels of suspended solids
- Small footprint
- Variety of screen sizes (15 to 500+ micron)
- Rapid backwash (~30 seconds)
- Steel or composite construction







Multi-Media Filtration Alternatives

(Continued)

Ultrafiltration (UF)

- Ideal for high levels of suspended solids
- Filtration of suspended solids to 0.2 microns
- Backwash based on pressure drop
- 95% recovery
- Chlorine-tolerant

8" X-Flow UF element (40 m2)



0.8 mm PES fibres pore size 30 nm



Corrugated plates: → optimal hydrodynamics

Bypass tubes: → minimum pressure loss in housing



Iron Removal Media Filters



Why Remove Iron?

- Iron easily falls out of solution and sticks to surfaces (coagulant)
- Coats plumbing, valves, equipment, boiler tubes, etc.
- Stains surfaces
- Fouls RO membranes and plugs filters

Iron Removal Media Filters



Birm / KDF[®] Filters

- Iron removal
- H₂S removal (rotten egg smell)
- Heavy metals removal
- May require pre-chlorination to oxidize high iron levels

Manganese Greensand Filters

- Iron removal
- H₂S removal
- Media regenerated with potassium permanganate
- Potassium permanganate feeder or continuously dosed
- Greensand Plus[®] does not require regeneration (KmNO₄)

Activated Carbon Filters



3/15/2013

Advantages

- Effective de-chlorination
- No chemicals required
- Low mechanical maintenance

Disadvantages

- Not always the best choice for organics removal
- Supports bacteria growth
- Difficult to sanitize chemically
- Sheds particles backwash as

infrequently as possible



Granulated Activated Carbon (GAC)







Water Softeners (Ion Exchange)





Twin Alternating Softener System

Single Softener System

Ion Exchange (Softening)



- When many compounds *dissolve* in water they break up into their component parts, which are *charged particles*
- These particles are atoms or groups of atoms called ions
- lons can be *positively charged* or *negatively charged*
- *Positive* charged ions are called *cations*: Sodium, Calcium and Magnesium
- *Negative* charged ions are call *anions:* Chloride, Sulfate, and Bicarbonate
- Molecules having one extra electron are called monovalent ions, those having two extra electrons are called *divalent* ions

Ion Exchange (Softening)



- Softening (reduces hardness)
- Hardness: Water that produces a scale forming & lather inhibiting tendency
- Utilizes resin to exchange undesirable dissolved ion for acceptable replacement ion
- Cation (softening) or anion removal
- Requires regeneration to restore effectiveness of resin
- Time between regenerations dependent on influent ion concentration

Ion Exchange (softening)



- Exchanges Ca++ (Calcium) and Mg++ (Magnesium) ions with Na+ (Sodium)
- Softening does not reduce TDS
- Regenerates with salt brine solution (NaCl)
- Brine disposal restricted in some areas
- Different types of resin available: Cation, anion, mixed bed...
- In high flow applications, regeneration costs (for salt) may be prohibitive

Water Softeners





- Multiple types (resins designed for specific applications)
- Modular design
- Optional skidmounted packages
- Low maintenance
- Easy to install
- FRP or steel tanks
- PLC control options

Chemical Dosing



Usually Consists of:

- Chemical solution tank
- Dosing pump (metering)
- Static mixer (may be required)
- Retention tank (may be required)





Chemical Dosing



Types:

- Anti-Scalant Dispersant that prevents scale build-up on membranes; Alternative to water softening; *More cost effective than softening on industrial applications*
- Acid Lowers pH
- Caustic Elevates pH (usually NaOH)
- Chlorine/Biocide To destroy organics
- **Sodium Metabisulfite** Chlorine removal; Alternative to carbon filtration; *More cost effective than carbon filter on industrial applications*

Reverse Osmosis Fundamentals



OSMOSIS

Water flows from low concentration of salts to higher concentration.

 $\pi_1 < \pi_2$

EQUILIBRIUM

Osmotic Pressure is the pressure required to stop water flow and reach equilibrium.

 $P = \Delta \pi$

REVERSE OSMOSIS

By applying pressure greater than Osmotic Pressure, flow of water is reversed; water flows from higher concentration solution to lower.

$$P > \Delta T$$

CRANE

Feed 100 gpm @ 500 PPM TDS Concentrate (waste) 25 gpm @ ~2,000 PPM

To drain

Feed (gpm) = Permeate (gpm) + concentrate (gpm)

- Rejection = % of salts (TDS) that do not pass through membrane Typically 99%
- Recovery = % of feed flow that passes through membrane (permeate) Maximum is typically 75 to 80%

Reverse Osmosis Terms



% Recovery

• The ratio of the permeate flow to the feed flow, usually expressed as a percentage

% Recovery =
$$\left[\begin{array}{c} Permeate Flow \\ Feed Flow \end{array}\right] \times 100$$

% Rejection

• The percentage of feed water TDS that does not pass through the membrane

% Rejection =
$$\begin{bmatrix} 1 - \frac{\text{Permeate TDS}}{\text{Feed TDS}} \end{bmatrix} \times 100$$

Flux

• Unit flow of water through a unit area of membrane per unit of time

F

- The most common units of measure are GFD (Gallons / Square Foot Day)
- The higher the flux, the faster membranes will foul which decreases system performance

$$lux (gfd) = \frac{Permeate Flow (gpd)}{Membrane Area (ft^2)}$$

Spiral Wound RO Elements





CE RO-Centric Design (continued)



- All systems designed for 3rd year of membrane life
- >40 PSI feed water pressure required <u>at entrance of RO skid</u> (be sure to account for pressure drops through pre-treatment tanks)
- Design for local electrical power supply (voltage/frequency/phase)
- Comply with membrane manufacturer's recommendations for:
 - Flux
 - Min/Max concentrate flow
 - Min/Max feed flow
 - Operating pressures

CE RO-Centric Design (continued)



Common Optional Features

- High TDS alarm / divert
- ORP meter to monitor / prevent chlorine damage
- Clean-In-Place (CIP) or cleaning station (CLS)
- Remote monitoring/controlling
- Fully containerized systems
- Chemical dosing monitoring/controlling









- Permeate Storage Tanks level controls, bulkhead fittings
- U/V Sterilization Lights To destroy organic and biological material
- Ozone Generation Systems Purification; eliminates the need to use chlorine; widely used at bottled water plants
- pH Adjustment Elevate or lower permeate pH through chemical dosing, calcite filters, etc.
- Chemical Dosing chlorination, pH adjustment
- Pump Skids Feed water, booster pumps, delivery pumps, etc.



Double / Triple Pass RO Systems

- System design links 2 to 3 RO systems in series
- May be integrated in single or multiple skid designs
- Used in applications requiring high purity water as in high pressure boilers
- Produces ultrapure water with < 1ppm (TDS) without the need for traditional DI tanks
- Does not require additional chemicals or regeneration
- Most economical (equipment & operational costs) means to produce ultrapure water
- DPRO[™] and TPRO[™] systems utilize concentrate recycle to achieve the same overall recovery as single-pass RO's



CE RO-Centric Design: Ultra-Pure Applications

Deionization (DI)

- Utilizes specially-engineered ion exchange resin to remove additional dissolved solids (cation / anion)
- Can be separate bed or mixed bed design
- Requires regeneration with strong caustic and acid solutions
- Waste from regeneration must be neutralized before disposal
- Regeneration is either performed in situ or through tank exchange programs
- Labor-intensive and high servicing costs



CE RO-Centric Design: Ultra-Pure Applications



Electrodeionization (EDI)

- Incorporate membrane / resin hybrid design using high voltage DC current as the driving force to reduce TDS to ultrapure levels
- lons removed continuously
- No regeneration chemicals
- Mixed bed resins for best silica removal
- Continuous flow of up to 18 meg-ohm water





Deaeration



Deaeration and Boiler Operation & Maintenance

- Commercial and Industrial Boilers operate at their best when the incoming boiler feed water is free from dissolved gasses
- Dissolved oxygen will cause severe corrosive damage to the boiler internals
- Dissolved oxygen also combines with any dissolved carbon dioxide to form carbonic acid which accelerates corrosion
- Deaerators are designed to remove oxygen down to 7 parts per billion (ppb) by weight (0.0005 cm³/L) or less
- A properly operated and well maintained deaerator will dramatically improve boiler lifespan and performance as well as reduce the volume of chemical oxygen scavenger required, such as sodium sulfite (Na₂SO₃)

Deaeration



Cochrane Counterflow Spray-Tray Deaerators



Vertical Configuration

Horizontal Configuration

Deaeration



Cochrane Parallel Downflow Spray-Tray Deaerators



Vertical Configuration

Horizontal Configuration



- Source water contains sediment and dissolved solids
- When this water is *evaporated (or boiled),* water leaves the system as vapor, *but the sediment and dissolved solids are left behind* in a lesser amount of water
- The result: the original dissolved solids and sediment are present at a higher *concentration*
- As the concentration of a dissolved solid *increases*, its tendency to *precipitate* (form a solid) from solution increases

Why treat boiler feed water?



- As the dissolved solids concentrate and begin to precipitate, they form scale on any available surface, i.e. heating surfaces in the boiler drum
- This scale acts as an inhibitor to heat transfer and the boiler will begin to operate below capacity and may eventually fail
- The concentration at which a dissolved solid will scale decreases as pressure increases; this indicates an increasing purity requirement with increasing boiler pressure
- <u>Greatly reduce Blowdown</u>

Boiler Water Chemistry



ABMA Boiler Drum Water Recommendations



Post R.O. Blowdown Calculation



Assumptions:	Average Steam Production = 300,000 pounds/hr
	Steam Pressure = 600 psig
	Chloride Concentration in Feed Water = 150 mg/l
	Current Boiler Blowdown rate = 12%
	Chloride Concentration in Feed Water treated by RO = 1 mg/l
	<i>Current Chloride Concentration in Drum = 1,000 mg/l</i>
	Desired Chloride Concentration in $Drum = 200 \text{ mg/l}$
	Boiler Efficiency = 82%
	Energy Costs = \$10 / million BTU
	Average Makeup Water Temperature = $40 ^{\circ}F$
Current Boiler Feedwa	$ster \ Usage = \frac{SteamOutput}{(1 - Blowdown)} = \frac{300,000 \ pph}{1 - 0.12} = 340,909 \ pph$
Boiler Blowdown Rate	using RO = $\frac{C_{feed}}{C_{boiler}} = \frac{1mg/l}{200mg/l} = 0.5\%$
Boiler Feedwater Usa	ge using $RO = \frac{SteamOutput}{(1 - Blowdown)} = \frac{300,000pph}{1 - 0.005} = 301,508pph$
Makeup Water Saving	s = 340,909 - 301,508 = 39,401 pph
Thermal Energy Savin	$gs = Enthalpy_{boilerwater} - Enthalpy_{feedwater} \approx 450BTU / lb$
Total Energy Savings	$=\frac{39,401pph \times 450BTU / lb \times \$10 / millionBTU}{0.82 \times 1,000,000} = \$216.22 / hr$
Annual Energy Saving	s = \$1.89 million

Annual Water Savings = 41,385,000 gallons (cost per gallon should include pumping costs, chemical treatment costs and any usage fees)



Field & Technical Services



Complete product-line field service support – worldwide...

- Commissioning / startup services
- Yearly service contracts
- System maintenance (weekly/monthly)
- Emergency technical support 24/7
- Remote monitoring / troubleshooting
- Ongoing operator training



EPRO Medium RO's





21,500 GPD



19,000 GPD

EPROi Large RO's





216,000 GPD



3/15/2013

EPROi Large RO's





EPROi SW Seawater RO's(17k – 40k+ ppm)





360,000 GPD (From Seawater)



EPROi SW Seawater RO's(17k-40k+ppm)



20,000 GPD (From Seawater Skid-mounted Pretreatment)



ERI PX System (State of the art energy recovery Technology)

Pretreatment





Multimedia Filters

(Skid-mounted with Booster Pumps)

Pretreatment





Softeners

(Skid-mounted SS Twin Alt. 60x67" Tanks)

Pretreatment





Carbon Filters (Skid-mounted FRP Triplex 48x72" Tanks)



Multimedia Filters (Skid-mounted FRP Duplex 30x72" Tanks)

Post Treatment





Electrodeionization (EDI) (Skid-mounted 25 GPM EDI 30 GPM RO)



Electrodeionization (EDI) (Skid-mounted 100 GPM EDI)

Post Treatment



Deionization (DI)

(Skid-mounted 45 GPM Mixed Bed)





EPROi Systems















Questions?







Local Contacts:

Frost Engineering Service Co.-NW, Snohomish, WA

- Eric Hamilton
- (503) 793-8068
- EricH@frostnw.com

Crane Environmental, Fallbrook, CA

- Rob Haney, Regional Sales Manager
- (760) 723-5587 O; (760) 468-3594 Cell
- <u>rhaney@cranewater.com</u>